

FIVE STAR CHEMICALS



Safety First!

Hazardous Charts

Health Hazard



- Carcinogen
- Mutagenicity
- Reproductive Toxicity
- Respiratory Sensitizer
- Target Organ Toxicity
- Aspiration Toxicity

Exclamation Mark



- Irritant (skin and eye)
- Skin Sensitizer
- Acute Toxicity (harmful)
- Narcotic Effects
- Respiratory Tract Irritant
- Hazardous to Ozone Layer (Non-Mandatory)

Flame Over Circle



- Oxidizers

Skull and Crossbones



- Acute Toxicity (fatal or toxic)

Corrosion



- Skin Corrosion/ Burns
- Eye Damage
- Corrosive to Metals

Flame



- Flammables
- Pyrophorics
- Self-Heating
- Emits Flammable Gas
- Self-Reactives
- Organic Peroxides

HAZARDOUS MATERIALS CLASSIFICATION



Safety First!

Personal Protective Equipment (PPE)

Personal Protective Equipment



Proper PPE: Safety Glasses with Side Shields, Arms and Legs covered with clothes, Rubber Boots, Chemical Resistant Gloves, Mask and Respirator

*Check you Safety Data Sheets



Safety Data Sheet SDS



Five Star Chemicals
& Supply, Inc.

Star San Safety Data Sheet

according to the Hazardous Products Regulation (February 11, 2015)

Issue date: 10-23-2020 Revision date: 11-17-2020 Version: 1.1

SECTION 1: Identification	
1.1. Product identifier	
Product form	: Mixture
Product name	: Star San
1.2. Recommended use and restrictions on use	
Recommended use	: Cleaner
1.3. Supplier	
Manufacturer	Distributor
Five Star Chemicals & Supply Inc	
5870 W. 52nd Ave	
Arvada, CO 80002	
T (303)287-0186	
1.4. Emergency telephone number	
Emergency number	: 800-535-5035 (Infotrac)
SECTION 2: Hazard identification	
2.1. Classification of the substance or mixture	
Classification (GHS CA)	
Skin Corr. 1B	H314
Eye Dam. 1	H318
HNOC	1
2.2. GHS Label elements, including precautionary statements	
GHS-CA labelling	
Hazard pictograms (GHS-CA)	:
Signal word (GHS CA)	: Danger
Hazard statements (GHS-CA)	: H314 - Causes severe skin burns and eye damage. Causes severe damage to the respiratory tract.
Precautionary statements (GHS-CA)	: P260 - Do not breathe dust/fume/gas/mist/vapours/spray. P264 - Wash hands thoroughly after handling. P280 - Wear eye protection, protective gloves. P301+P330+P331 - IF SWALLOWED: Rinse mouth. Do NOT induce vomiting. P303+P361+P353 - IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water. P363 - Wash contaminated clothing before reuse. P304+P340 - IF INHALED: Remove person to fresh air and keep comfortable for breathing. P305+P351+P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P310 - Immediately call a POISON CENTER or doctor. P405 - Store locked up. P501 - Dispose of contents/container to hazardous or special waste collection point, in accordance with local, regional, national and/or international regulation.
2.3. Other hazards	
No additional information available	
2.4. Unknown acute toxicity (GHS CA)	
Not applicable	
SECTION 3: Composition/information on ingredients	
3.1. Substances	
Not applicable	

1. Identification
2. Hazard(s) Identification
3. Composition/Ingredients
4. First-aid Measures
5. Firefighting Measures
6. Accidental Release Measures
7. Handling & Storage
8. Exposure/Personal Protection

9. Physical & Chemical Properties
10. Stability & Reactivity
11. Toxicological Information
12. Environmental Information
13. Disposal Considerations
14. Transportation Information
15. Regulatory Information
16. Other Information



The Three-Step Cleaning Process

1

Alkaline & Caustic Cleaners

Sodium/ Potassium Hydroxide
Sodium Carbonate / Metasilicate
Chlorinated

Remove organic matter;
proteins, starches and
carbohydrates

**PBW, HD#2, CMC, LLC,
Super CIP**

2

Acid Rinse / Wash

Nitric Phosphoric Acid Blend

Remove and helps prevent
beer stones / water scale,
fats, oils

**Acid #5, Phosphoric Acid,
Citric Acid**

3

Sanitizer / Final Rinse

PAA, Phosphoric Acid,
Iodaphor, Chlorine Dioxide
Isopropyl Alcohol

Reduce Microbiological
and Bacteria to 99.9%

**Star San, SaniClean, PAA Pro,
Star-Xene**



Caustic and Alkaline Removing Organic Material

Standard
Soil

Medium
Soil

Heavy
Soil

*Use in Moderation

Alkaline

**Non-Chlorinated
Caustic**

**Chlorinated
Caustic**

Sodium Carbonate / Percarbonate /
Metasilicate

Sodium Hydroxide
Hydrogen Peroxide booster

Sodium Hypochlorite

**Gentler on the
stainless steel**

**More aggressive on the
stainless steel**

**Very aggressive on the
stainless steel**

**Reduced risk of corrosion
Can be used on “soft” metal**

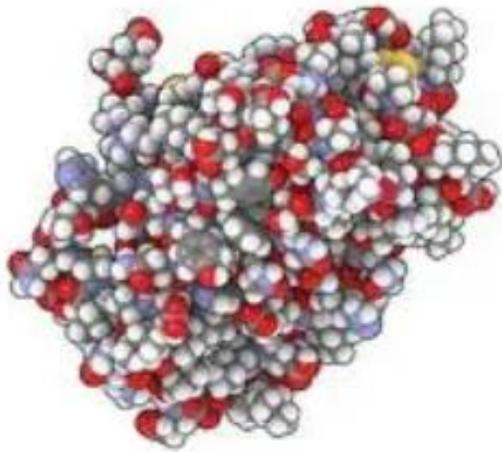
**Can set beer stones
Can be used on “soft” metal
But NOT at Higher temps**

**Need to complete rinse
Can cause pitting
Can't be used on “soft” metal**

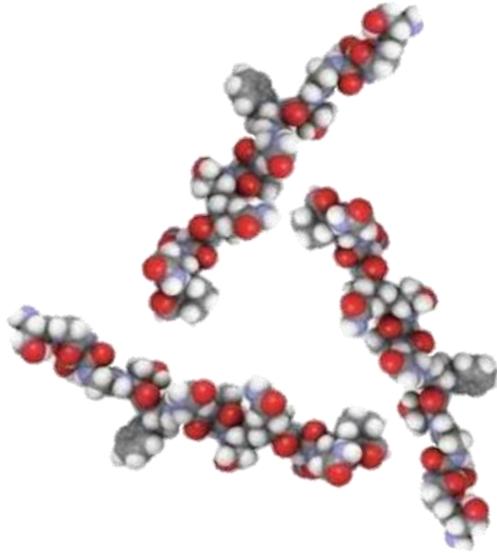


Caustic and Alkaline

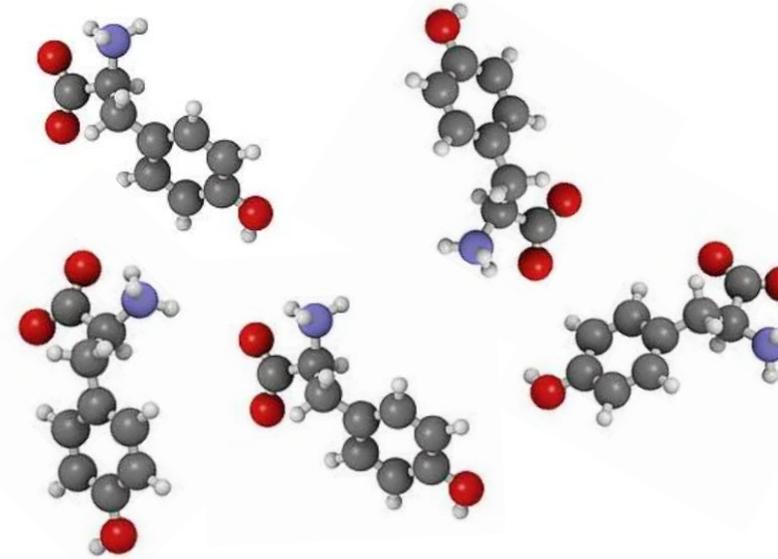
Protein



Peptide



Amino Acid



Hydrolytic: Breaks bonds protein by replacing with water molecule

Phosphoric Acid and Nitric Acid

Phosphoric Acid

Removes Mineral Deposits (Beer Stones)

Removes Rust

Neutralizes Residual Alkalis

Preps Surface for Passivation

*Can be Biocidal but in high concentrations

Nitric Acid

Removes Mineral Deposits (Beer Stones)

Biocidal Properties (Kills Bacteria & Fungi)

Dissolves Free Floating Iron Ions (Passivation)

Nitric Acid & Phosphoric Acid Blends

More Nitric than Phosphoric= used for Acid Rinse, Acid Wash and Passivation

Product: Acid #5

More Phosphoric than Nitric= Keg Cleaning, Hard to remove Scale/Beer Stones

*Some Contain Detergents

*Some can be used in CO2 environments

Product: Acid #6

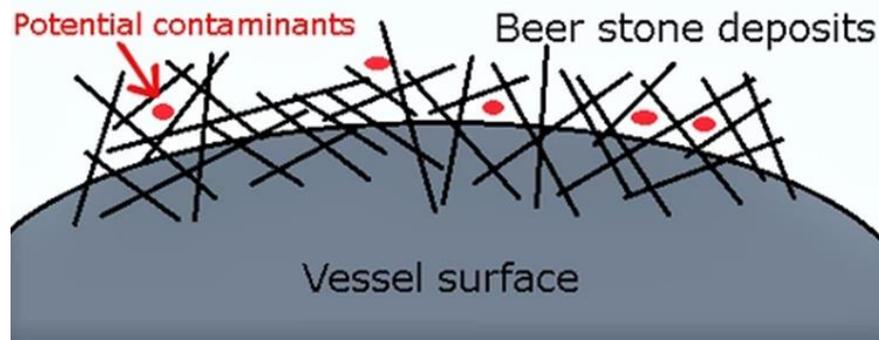
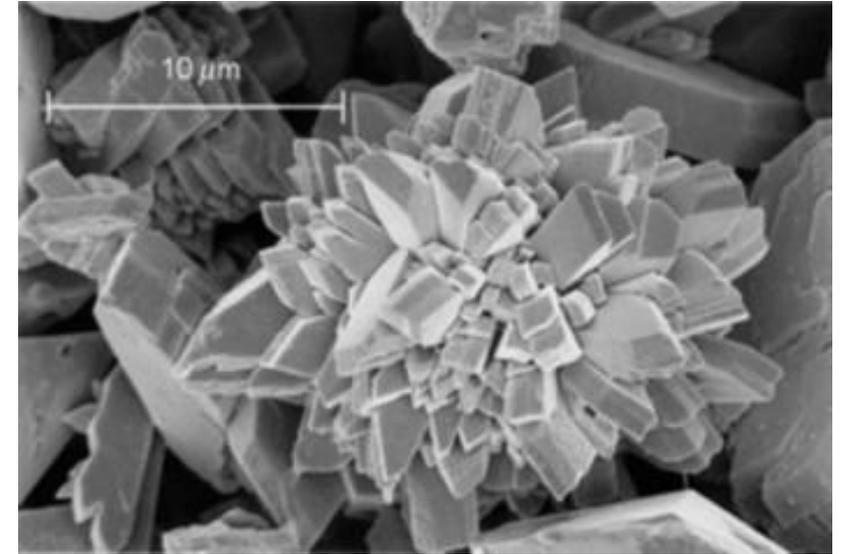


Beer Stones

What is a Beer Stone?: Calcium Oxalate

Calcium Oxalate: Crystallized compound that forms when calcium and magnesium carbonate binds with proteins

Why is this Bad?: Harbors bacteria that can cause off flavor, spoil beer, create a cloudy appearance



Bonus: Where else do you find Calcium Oxalate?



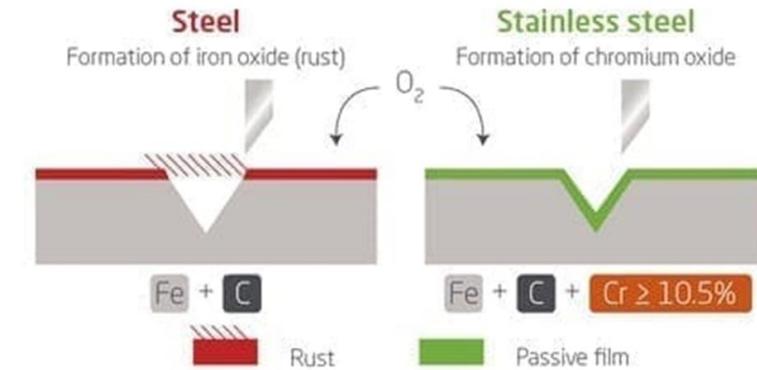
How to Prevent Scale & Beer Stone Build Up

Passivation: A chemical process that makes Stainless Steel and Copper surfaces less reactive and more resistant to corrosion by forming a protective layer

Stainless Steel Composition: 10.5% Chromium & 50-70% Iron

Iron and Oxygen form Iron Oxide= Rust

Chromium and Oxygen form Chromium Oxide= Passivation Layer

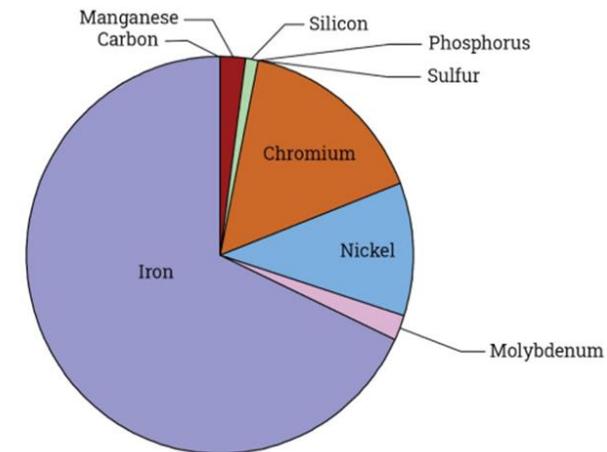


Nitric Acid Phosphoric Acid Blend

Phosphoric Acid: Preps the surface for by dissolving mineral deposits & removing containments

Nitric Acid: Removes the free-floating iron ions

Allows the Chromium and the oxygen to interact creating the protective layer



Methods of Passivation

Method One:

1. CIP with an Alkaline cleaner or Caustic Cleaner
2. Nitric Phosphoric Acid blend (needs more nitric than phosphoric)
 - a) Do NOT rinse
 - b) Allow to air dry for 24 hours

Method Two:

1. CIP with an Alkaline cleaner or Caustic Cleaner
2. Nitric Phosphoric Acid blend (needs more nitric than phosphoric)
 - a) Drain but do NOT rinse
3. Follow with an Alkaline cleaner CIP
 - a) **MUST CONTAIN** hydrogen peroxide or sodium percarbonate
4. Rinse and allow to air dry for 1-2 hours

*This method forms a **phosphate/silicate conversion coating** that protects the surface and creates a glassy-smooth finish



Methods of Passivation

Method Three:

1. CIP with an Alkaline cleaner or Caustic Cleaner
2. Citric Acid Bath
3. Drain and Rinse
4. Allow to air dry for 24 hours

Citric Acid

Can be used prior to passivating with Phos/Nitric to remove rust



Cleaning Best Practices

Scheduled Maintenance

Create a schedule for regularly cleaning and complete disassembly

Label

When taking apart equipment, label the parts. Take photos or video tape throughout the process if needed.

Disassemble Faucets / Clamps

Take apart and clean all faucets and clamps. Replace any damaged seals or gaskets

Use appropriate tools:

Avoid using abrasive scrubbers on stainless steel, as they can create scratches.

Use color-coded and labeled tools for different areas to prevent cross-contamination.



Questions?



Thank You

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